Studies on Chirality Transfer in the [2,3] Sigmatropic Rearrangement

of Anions Derived from Trialkylstannylmethyl Allylic Ethers.

Stereospecific Synthesis of (2R)-3-Benzyloxy-2-methylpropanol.

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Abstract: Virtually complete chirality transfer is observed in the [2,3] sigmatropic rearrangement of the anion derived from trialkyl-stanylmethyl (E)- or (Z)-allylic ethers.

The recent focus of attention on the [2,3] sigmatropic rearrangement has substantially reinforced the arsenal of stereocontrol for acyclic systems.¹ The scope of the [2,3] sigmatropic (Wittig) rearrangement was considerably expanded by Still's discovery that anions derived from transmetallation of trialkylstannylmethyl allylic ethers can undergo a facile rearrangement. Still examined the olefin geometry of the resulting homoallylic alcohols.² Recent attention has focused on the stereochemistry of the migrating termini as well as the olefin geometry. Substitution patterns at the anion center as well as on the allylic ether can have profound effects on the stereochemistry of the rearranged product. It was not clear what effect would be exhibited by the parent anion. Because of the synthetic potential of these transformations³ we have explored the stereochemistry of this reaction. Herein we report our results and the application of the reaction for the preparation of (2R)-3-benzyloxy-2methylpropanol.

Our starting material 1 (92% e.e. by analysis with NMR shift reagent) was obtained by the reduction of 5-methyl-2-hexyn-4-one with <u>B</u>-3-pinanyl-9-BBN (R-Alpine Borane, from 92% e.e. $(+)-\alpha$ -pinene).4 (<u>Scheme I</u>). Partial hydrogenation of 1 using Lindlar catalyst provided (<u>Z</u>)-allylic alcohol 2 (92% e.e.), which was then deprotonated with NaH and the resulting alkoxide reacted with trimethylstannylmethyl iodide to afford ether 3. Treatment of 3 with <u>n</u>-BuLi at -78° followed by warming to 0° provided only (<u>E</u>)-homoallylic alcohol 4, [a]p + 28.4 (<u>c</u> 1, CHCl₃). The enantiomeric purity of 4 was determined to be 92% by NMR shift studies (Eu(hfc)₃). Since the starting material was also 92% e.e. the reaction proceeds with complete chirality transfer. The absolute configuration of 4 was verified as follows. Ozonolysis of benzyl-protected 4 and subsequent reduction of the ozonide provided $(2\underline{R})$ -3-benzyloxy-2-methylpropanol⁵, $[\alpha]_D$ -3.39, (\underline{c} 1.68, EtOH), 1it.,^{5a} $[\alpha]_D$ -4 (\underline{c} 1.076, EtOH), and $[\alpha]_D$ +10.7 (\underline{c} 3.10, CHCl₃), 1it.^{5b} $[\alpha]_D$ +11.2 (\underline{c} 16, CHCl₃). This route not only verifies the absolute configuration of 4, but also represents an asymmetric synthesis of (2\underline{R})-3-benzyloxy-2-methylpropanol (5) which has been commonly produced by microbial oxidation of isobutyric acid with <u>pseudomonas putida</u>.⁵ This chiral unit has been extensively used in the synthesis of more complex molecules.

In the case of the (<u>E</u>)-allylic system (readily available by LAH reduction of 1) treatment of ether (7) with <u>n</u>-BuLi afforded a mixture of (<u>E</u>)- and (<u>Z</u>)-homoallylic alcohols (8 and 9) in a 53:47 ratio. The two products (8 and 9) were isolated by HPLC. The optical rotation of 8 was of the same magnitude but opposite sign of 4 ($[\alpha]_D$ -28.1 (<u>c</u> 1, CHCl₃)). Analysis by NMR (Eu(hfc)₃) indicated a 92% e.e.. Hence, the absolute configuration of 8 and its enantiomeric excess were confirmed. Analysis of 9 by NMR indicated a 92% e.e. of the opposite configuration at C-2. Hydrogenation of 8 and 9 provided 10 and 11 respectively. (10, $[\alpha]_D$ -3.75 (<u>c</u> 0.8, CHCl₃) and 11 $[\alpha]_D$ +3.64 (c 0.77, CHCl₃)).

The stereochemistry and selectivity of the transformation may be rationalized by consideration of the possible 5-membered ring transition states (<u>Scheme II</u>). In the case of isomerization of the <u>R,Z</u> isomer only the transition state leading to the <u>R,E</u> isomer (4) is favorable. The transition state leading from the <u>R,Z</u> starting material to the <u>S,Z</u> product is destabilized by the methyl-isopropyl interaction. However, in the case of the <u>R,E</u> isomer, neither transition state is destabilized by the methyl-isopropyl interaction. Of course one must be cautioned that these pictures are greatly oversimplified. For example, they do not take into account the nature of the carbanion or the position of the counter ion.

In conclusion, the rearrangement of these anions is stereospecific. However, the stereochemistry of the new chiral center is intimately related to the geometry of the olefinic product. The \underline{Z} starting



(a) H₂, Lindlar cat. 98%. (b) 1.4 eq. NaH, 1 eq. ICH_2SnMe_3 , THF reflux, 17 h, 75%. (c) 2 eq. n-BuLi, -78°, 40 min, 0°, 40 min, 72% (d) 1.8 eq. NaH, 2.2 eq. benzyl chloride, THF, reflx, 15 h, 87%. (e) Ozone, -78°, NaBH₄, -78° + RT, 15 h, 78%. (f) 8 eq. LAH, THF, reflux, 4 h, 74%. (g) H₂, 5% Pd/C, 100%.

Scheme II









material provides a single product which may be converted into the chiral unit 5. Either enantiomer of 5 may be prepared by the appropriate asymmetric reduction of the starting propargyl ketone.

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